# MAASAI MARA UNIVERSITY 

# REGULAR UNIVERSITY EXAMINATIONS 2018/2019 ACADEMIC YEAR FOURTH YEAR, SECOND SEMESTER 

SCHOOL OF SCIENCE BACHELOR OF SCIENCE CHEMISTRY

## COURSE CODE: CHE 417

COURSE TITLE: BIOINORGANIC CHEMISTRY

DATE: $18^{\text {TH }}$ APRIL 2019
TIME: 0830-1030HRS
INSTRUCTIONS TO CANDIDATES

1. Answer Question ONE and any other TWO questions
2. No writing on the Question paper

## Question One (30 marks)

(a) Define /explain the following
i. Isomerism
ii. Porphyrins
iii. Hemethrins
iv. Metalloprotein
v. Cooperativity
vi. Cytochrome
(b) Explain the functions of hemoglobin
(c) Explain the term lanthanide contraction
(d) Draw the structure of corrin and state one of its functions
(e) Giving at least two examples distinguish between hard and bases soft
(f) Explain why lanthanides especially gadolinium +3 ions make a good agent MRI applications
(3 marks)

## Question Two (20 marks)

(a) (i) Draw a curve to explain the variations in oxygen affinity to myoglobin and hemoglobin
(ii) Explain the cooperative binding of hemoglobin
(b) Briefly present the aqueous iron chemistry in relation to the mineralization
(c) Give reasons why organisms mineralize iron

## Question Three ( 20 marks)

(a) Draw the structure porphyrin ring
(b) List three functions of Coenzyme- $\mathrm{B}_{12}$
(c) Describe the process of nitrogen fixation
(d) Explain the iron storage protein ferritin

## Question Four (20 marks)

(a) Discuss the process of photosynthesis
(10 marks)
(b) Explain transport, formation and degradation of hydrogen carbonate in our body
(10 marks)
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