

MAASAI MARA UNIVERSITY

REGULAR UNIVERSITY EXAMINATIONS 2018/2019 ACADEMIC YEAR FOURTH YEAR, SECOND SEMESTER

SCHOOL OF SCIENCE BACHELOR OF SCIENCE CHEMISTRY

COURSE CODE: CHE 417

COURSE TITLE: BIOINORGANIC CHEMISTRY

DATE: 18TH APRIL 2019

TIME: 0830 - 1030HRS

INSTRUCTIONS TO CANDIDATES

- 1. Answer Question **ONE** and any other **TWO** questions
- 2. No writing on the Question paper

Question One (30 marks)

(a) Define /explain the following	(12 marks)
i. Isomerism	
ii. Porphyrins	
iii. Hemethrins	
iv. Metalloprotein	
v. Cooperativity	
vi. Cytochrome	
(b) Explain the functions of hemoglobin	(6 marks)
(c) Explain the term lanthanide contraction	(2 marks)
(d) Draw the structure of corrin and state one of its functions	(4 marks)
(e) Giving at least two examples distinguish between hard and bases soft	(3 marks)
(f) Explain why lanthanides especially gadolinium +3 ions make a good agent MRI	
applications	(3 marks)
Question Two (20 marks)	
(a) (i) Draw a curve to explain the variations in oxygen affinity to myoglobin and hemoglobin	
	(5 marks)
(ii) Explain the cooperative binding of hemoglobin	(5 marks)
(b) Briefly present the aqueous iron chemistry in relation to the mineralization	(5 marks)
(c) Give reasons why organisms mineralize iron	(5 marks)
Question Three (20 marks)	
(a) Draw the structure porphyrin ring	(2 marks)
(b) List three functions of Coenzyme-B ₁₂	(3marks)
(c) Describe the process of nitrogen fixation	(10 marks)
(d) Explain the iron storage protein ferritin	(5 marks)
Question Four (20 marks)	
(a) Discuss the process of photosynthesis	(10 marks)
(b) Explain transport, formation and degradation of hydrogen carbonate in our body (10 morks)	
//END	(10 marks)